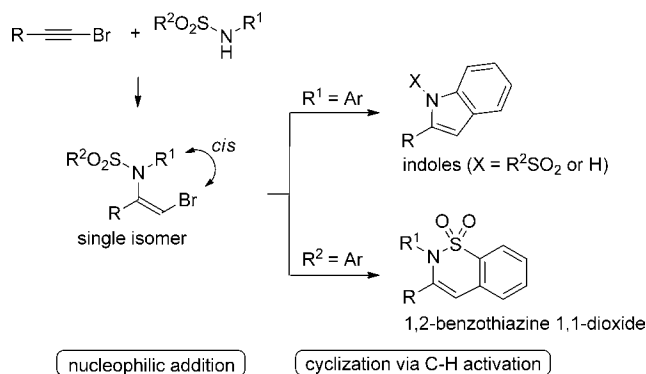


Facile Preparation of Indoles and 1,2-Benzothiazine 1,1-Dioxides: Nucleophilic Addition of Sulfonamides to Bromoacetylenes and Subsequent Palladium-Catalyzed Cyclization**

Masahito Yamagishi, Ken Nishigai, Azusa Ishii, Takeshi Hata, and Hirokazu Urabe*

Nucleophilic addition to an electron-deficient carbon–carbon multiple bond is one of the most fundamental reactions in organic chemistry.^[1] Among the various electron-withdrawing groups adjacent to an olefin or acetylene, thus making this process feasible, the weakest possible substituents, such as halogens, have recently attracted increasing attention.^[2–7] We have already reported that certain nitrogen-containing compounds such as *N*-alkylsulfonamides,^[8] imidazolines, or imidazoles undergo nucleophilic addition to 1-halo-1-alkynes.^[5d,e] Furthermore, the synthetic utility of the adducts obtained by this reaction has been explored in a cyclization reaction affording nitrogen heterocycles.^[5d] Our continuing efforts along these lines allow us to report herein, a novel entry to indole synthesis,^[9] which consists of the nucleophilic addition of *N*-arylsulfonamides (i.e. *N*-sulfonylanilines) to bromoacetylenes with subsequent palladium-catalyzed cyclization of the resultant *cis*-2-(*N*-aryl-*N*-sulfonylamino)-1-bromoalkenes through aromatic C–H bond activation^[10] (Scheme 1). A new synthesis of another interesting heterocycle, 1,2-benzothiazine 1,1-dioxide,^[11] by a similar sequence is also described in this report.



Scheme 1. Facile preparation of two types of heterocycles by a novel synthetic strategy.

The nucleophilic addition of *N*-phenylmethanesulfonamide (**2a**) to 1-bromo-1-octyne (**1a**) proceeded under the same reaction conditions as reported previously for *N*-alkylsulfonamides,^[5d] and gratifyingly the yield of the adduct **3aa** (85 %, entry 1, Table 1) was higher than those in the

Table 1: Fundamental data for nucleophilic addition.

$\text{C}_6\text{H}_{13}\text{—Br} + \text{Ms—N(Ph)—H} \xrightarrow[\text{DMF, 120 } ^\circ\text{C}]{\text{K}_3\text{PO}_4 \text{ (1.5 equiv)}} \text{C}_6\text{H}_{13}\text{—N(Ph)—Ms—Br}$				
	1a (1 equiv)	2a	3aa single isomer	
Entry	2a (equiv)	<i>t</i> [h]	Yield [%] ^[a]	Recovery of 2a [%] ^[b]
1	3	6	85 (93) ^[c]	97
2	2	8	71	—

[a] Yield of isolated product is based on **1a**. [b] Yield of isolated product based on unreacted **2a**. [c] Yield based on consumed **2a**. DMF = *N,N*-dimethylformamide, Ms = methanesulfonyl.

previous cases (up to 78 %). The reaction was highly regio- and stereoselective such that other isomers were not detected in the crude reaction mixture. Although an excess amount of the sulfonamide **2a** (3 equiv relative to **1a**) guarantees a good product yield, it could be reduced to 2 equivalents and still keep the yield within an acceptable range (entry 2). When 3 equivalents of **2a** was used, the unreacted **2a** was recovered in 97 % yield and could be recycled; the yield of **3aa** based on the consumed sulfonamide **2a** was calculated to be 93 % (entry 1).

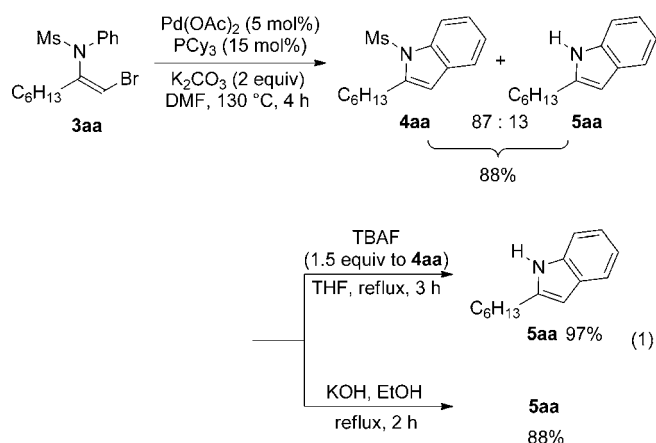
The *cis* relationship between the bromide and sulfonylamine moieties in the adduct **3aa** appears suitable for a variety of cyclizations,^[12] one of which is the palladium-catalyzed ring closure through activation of the aromatic C–H bond as illustrated in Equation (1). In fact, treatment of **3aa** with Pd(OAc)₂^[10] effected the cyclization to give the *N*-sulfonyl-indole **4aa** contaminated with a small amount of desulfonylated **5aa** in 88 % yield upon isolation. Despite considerable effort, the simultaneous production of **5aa** was hard to suppress and the ratio of **4aa** to **5aa** changed slightly from run to run. While **4aa** and **5aa** were separable by HPLC,^[13] the mixture was subjected to either desulfonylation with TBAF (Bu₄NF)^[14] or 10 % ethanolic KOH,^[15] thus leading to indole **5aa** in either 97 or 88 % yield.^[16]

Table 2 summarizes other results of the indole synthesis. In this sequence, the mixture of cyclization products **4** and **5** was directly submitted to the subsequent desulfonylation with TBAF. The yields of **4** and **5**, listed for reference, were estimated by ¹H NMR spectroscopy using an internal stan-

[*] M. Yamagishi, K. Nishigai, A. Ishii, Dr. T. Hata, Prof. Dr. H. Urabe
Department of Biomolecular Engineering, Graduate School of
Bioscience and Biotechnology, Tokyo Institute of Technology
4259-B-59 Nagatsuta-cho, Midori-ku, Yokohama
Kanagawa 226-8501 (Japan)
E-mail: hurabe@bio.titech.ac.jp
Homepage: <http://www.urabe-lab.bio.titech.ac.jp>

[**] This work was supported by a Grant-in-Aid for Challenging
Exploratory Research (22655014) from the JSPS (Japan).

Supporting information for this article is available on the WWW
under <http://dx.doi.org/10.1002/anie.201201024>.



dard to determine the amount of TBAF needed to insure the desulfonation step; typically 2 equivalents of TBAF were used relative to **4**. As far as the nucleophilic addition of *N*-phenylmethanesulfonamide (**2a**) to various bromoacetylenes (**1b–e**) is concerned, the corresponding adducts **3ba–ea** were obtained in good yields (entries 2–5). The mild reaction conditions allow the presence of representative functional groups, even an unprotected hydroxy group, on the side chain of bromoacetylenes. Likewise, 1-bromo-1-octyne (**1a**) and a variety of aniline derivatives, **2b–g**, afforded the desired products **3ab–ag** in good yields. The successful preparation of

the adducts **3aa**, **3ba–ea**, and **3ab–ag** prompted us to investigate the generality of the subsequent cyclization through an intramolecular C–H activation (**3**→**4**+**5**). As previously described for the preparation of **5aa** from **3aa** [Eq. (1)], all reactions inevitably produced a mixture of sulfonfylated indole **4** and a small amount of its parent indole **5**. The crude reaction mixtures were converted into indoles (**5aa**, **5ba–ea**, and **5ab–ag**) through desulfonation with TBAF in good overall yields (entries 1–11). As a whole, these reaction conditions were compatible with various functional groups both on the acetylene side chain and aromatic ring, thus demonstrating broad applicability.^[17]

In Table 2, we separately listed each result for the nucleophilic addition and the subsequent cyclization/desulfonation. However, more practically, these steps can be carried out in one pot without isolation of the intermediate products. Scheme 2 shows such an example in which indole **5aa** was obtained from the primary starting materials **1a** and **2a** in an excellent yield.

While the synthesis of unprotected indoles from *N*-arylmethanesulfonamides was established as above, sulfonyl-protected indoles could be cleanly produced from more robust *N*-aryl-*p*-toluenesulfonamides **6** as shown in Table 3. The nucleophilic addition of *N*-(*p*-toluenesulfonyl)aniline (**6a**) to bromoacetylenes **1a–e** (entries 1–5) required a longer reaction period (12 h) than that of the corresponding

methanesulfonamides (6 h, see Table 2), but the product yields remained at a similar level. The longer reaction time might be due to the enhanced steric hindrance of **6**. The subsequent palladium-catalyzed cyclization of the adducts **7aa–ea** and **7ab** proceeded in the same way as that of methanesulfonamides to give sulfonylindoles **8aa–ea** and **8ab**, respectively, however the desulfonlated products were not detected in all cases (entries 1–6, Table 3). It should be emphasized that the palladium-catalyzed cyclization exclusively occurred on the *N*-aryl group in preference to the *N*-tosyl group, thus producing only *N*-sulfonylindoles **8**.

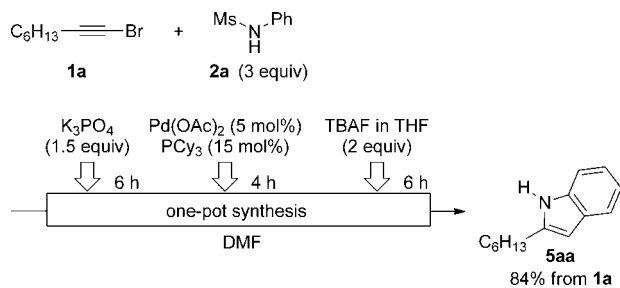
In conjunction with the indole synthesis discussed above, we noted in the preceding paragraph that the palladium-catalyzed cyclization of **7aa–ea** and **7ab** exclusively occurred at the aniline moiety to give **8aa–ea** and **8ab**, respectively, and the aromatic ring in the *p*-toluenesulfonyl group remained untouched. However, if the aniline moiety is not present in the molecule, then a similar cyclization did proceed toward the *p*-toluenesulfonyl group, thus leading to a new

Table 2: Preparation of indoles **5** from **1** and **2**.^[a]

$\text{R}-\text{C}\equiv\text{C}-\text{Br} + \text{Ms}-\text{N}(\text{H})-\text{Ar} \xrightarrow{\text{a}} \text{Ms}-\text{N}(\text{Ar})-\text{C}(\text{R})=\text{CH}-\text{Br}$ 1 2 3 single isomer					
$\xrightarrow{\text{b}} \left(\text{Ms}-\text{N}(\text{Ar})-\text{Indole}(\text{X}) + \text{H}-\text{N}(\text{Ar})-\text{Indole}(\text{X}) \right) \xrightarrow{\text{c}} \text{pure } \mathbf{5}$ 4 5					
Entry	R	Ar	Yield [%] ^[b]	Yield [%] ^[c] 4/5	Yield [%] ^[d]
1	C ₆ H ₁₃	(1a) Ph (2a)	3aa : 85	94/6	5aa : 87
2		(1b) Ph (2a)	3ba : 92	89/11	5ba : 84
3		(1c) Ph (2a)	3ca : 87	83/17	5ca : 76
4		(1d) Ph (2a)	3da : 64	53/47	5da : 79
5		(1e) Ph (2a)	3ea : 82	44/23	5ea : 65
6	C ₆ H ₁₃	(1a) <i>p</i> -MeC ₆ H ₄ (2b)	3ab : 79	90/10	5ab : 80
7	C ₆ H ₁₃	(1a) 3,5-Me ₂ C ₆ H ₃ (2c)	3ac : 85 ^[e]	89/8	5ac : 78
8	C ₆ H ₁₃	(1a) <i>p</i> -MeOC ₆ H ₄ (2d)	3ad : 92	89/11	5ad : 83
9	C ₆ H ₁₃	(1a) <i>p</i> -(Me ₂ N)C ₆ H ₄ (2e)	3ae : 92 ^[f]	95/5	5ae : 82
10	C ₆ H ₁₃	(1a) <i>p</i> -(AcHN)C ₆ H ₄ (2f)	3af : 87	89/10	5af : 92
11	C ₆ H ₁₃	(1a) <i>p</i> -ClC ₆ H ₄ (2g)	3ag : 78 ^[g]	78/18	5ag : 85 ^[h]

[a] Reaction conditions: a) Bromoacetylene **1** (1 equiv), sulfonamide **2** (3 equiv), K₃PO₄ (1.5 equiv), DMF, 120 °C, 6 h; b) Pd(OAc)₂ (5 mol%), PCy₃ (15 mol%), K₂CO₃ (2 equiv); DMF, 130 °C, 4 h; crude reaction mixture was directly used in the next step; c) TBAF (2 equiv to **4**); THF, reflux, 4.5–5.5 h.

[b] Yield of isolated product based on **1**. [c] Yield of crude product before desulfonation and determined by ¹H NMR spectroscopy with an internal standard. These data are shown only for reference. [d] Overall yield of the isolated product from **3**. [e] The reaction period was extended to 8 h. [f] Unreacted **2e** was recovered in 97% yield upon isolation and may be recycled. Thus, the yield of **3ae** based on consumed **2e** was 94%. [g] The reaction period was extended to 12 h. [h] The cyclization was performed for 1 h. Bn = benzyl.



Scheme 2. One-pot synthesis of indole **5aa** from **1a** and **2a**. THF = tetrahydrofuran.

Table 3: Preparation of *N*-sulfonylindoles **8** from **1** and **6**.^[a]

Entry	R	Ar	Yield [%] ^[b]	Yield [%] ^[c]
1	C ₆ H ₁₃	(1a) Ph (6a)	7aa : 80	8aa : 83
2		(1b) Ph (6a)	7ba : 72	8ba : 77
3	BnO-	(1c) Ph (6a)	7ca : 70	8ca : 89
4		(1d) Ph (6a)	7da : 69	8da : 88
5	HO-	(1e) Ph (6a)	7ea : 81	8ea : 60
6	C ₆ H ₁₃	(1a) 3,5-Me ₂ C ₆ H ₃ (6b)	7ab : 72 ^[d]	8ab : 70

[a] Reaction conditions: a) Bromoalkyne **1** (1 equiv), sulfonamide **6** (3 equiv), K₃PO₄ (1.5 equiv); DMF, 120 °C, 12 h; b) Pd(OAc)₂ (5 mol%), PCy₃ (15 mol%), K₂CO₃ (2 equiv); DMF, 130 °C, 4 h. [b] Yield of isolated product based on **1**. [c] Yield of isolated product from **7**. [d] The reaction period was extended to 14 h. Ts = 4-toluenesulfonyl.

preparation of 1,2-benzothiazine 1,1-dioxides as shown in Equation (2) (DMA = *N,N*-dimethylacetamide). For example, the *N*-alkyl-*p*-toluenesulfonamide **9a** underwent the nucleophilic addition to the bromoalkyne **1a** as described before^[5d] to give exclusively the *Z*-bromoalkene **10aa**. Its cyclization under the same reaction conditions as those in Table 3 gave 1,2-benzothiazine 1,1-dioxide **11aa** in 61 % yield, which was increased to 77 % under a new set of optimized reaction conditions. As expected from the fact that the cyclization to the *p*-toluenesulfonyl group is a less favorable path, the optimized reaction conditions required a larger amount of the palladium catalyst (10 mol %) relative to the original indole synthesis (5 mol %) to attain comparable product yields.

Results similar to those shown in Equation (2) are summarized in Table 4. 1-Bromo-1-octyne (**1a**) and *N*-alkyl-sulfonamides **9** derived from various arenesulfonyl groups afforded the corresponding *Z*-bromoalkenes **10aa–af** in good yields (entries 1–6). These products were successfully transformed into 1,2-benzothiazine 1,1-dioxides **11aa–af** (entries 1–6), wherein the cyclizations of **10ad** and **10ae**, having an electron-deficient aromatic ring (entries 4 and 5), were complete in shorter reaction periods than that of the others. Bromoalkene **10af**, having an unsymmetrical aromatic ring, cyclized to **11af** with a good regioselectivity (entry 6).

In conclusion, the nucleophilic addition of various sulfonamides to 1-bromo-1-alkynes took place in a highly regio- and

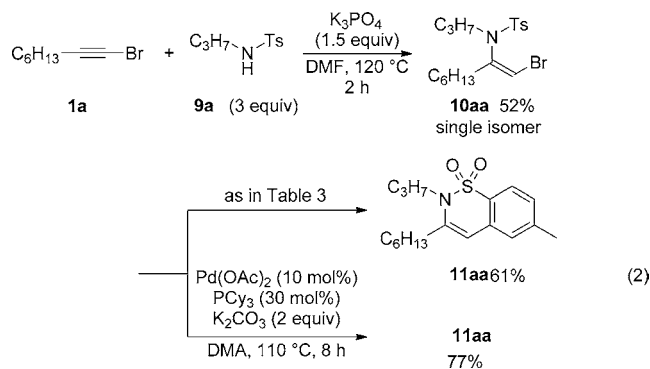
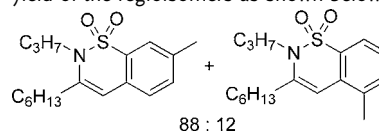


Table 4: Preparation of 1,2-benzothiazine 1,1-dioxides **11** from **1a** and **9**.^[a]

Entry	Ar	Yield [%] ^[b]	Yield [%] ^[c]
1	<i>p</i> -MeC ₆ H ₄ (9a)	10aa : 52	11aa : 77
2	Ph (9b)	10ab : 52	11ab : 71
3	<i>p</i> -MeOC ₆ H ₄ (9c)	10ac : 48	11ac : 73
4	<i>p</i> -NO ₂ C ₆ H ₄ (9d)	10ad : 53	11ad : 77 ^[d]
5	<i>p</i> -ClC ₆ H ₄ (9e)	10ae : 61	11ae : 58 ^[e]
6	<i>m</i> -MeC ₆ H ₄ (9f)	10af : 51	11af : 66 ^[f]

[a] Reaction conditions: a) Bromoalkyne **1a** (1 equiv), sulfonamide **9** (3 equiv), K₃PO₄ (1.5 equiv); DMF, 120 °C, 2 h; b) Pd(OAc)₂ (10 mol%), PCy₃ (30 mol%), K₂CO₃ (2 equiv); DMA, 110 °C, 8 h. [b] Yield of isolated product based on **1a**. [c] Yield of isolated product from **10**. [d] The reaction period was 2 h. [e] The reaction period was 3.5 h. [f] Combined yield of the regioisomers as shown below.



stereoselective manner to give (*Z*)-2-(sulfonylamino)-1-bromoalkenes in good yields. The *cis* structure of sulfonylamino and bromo substituents in the products should be quite useful for a variety of cyclizations, as demonstrated by the palladium-catalyzed cyclization involving aromatic C–H bond activation, thus providing a new and facile preparation of indoles, *N*-sulfonylindoles, and 1,2-benzothiazine 1,1-dioxides.

Received: February 7, 2012

Published online: May 13, 2012

Keywords: alkynes · cyclizations · indoles · nucleophilic addition · palladium

- [1] a) F. A. Carey, R. J. Sundberg, *Advanced Organic Chemistry, 5th ed., Part B*, Springer, New York, 2007, pp. 183–200; b) M. B. Smith, J. March, *March's Advanced Organic Chemistry, 6th ed.*, Wiley, Hoboken, 2007, pp. 1130–1132; c) *Comprehensive*

- Organic Synthesis, Vol. 4* (Eds.: B. M. Trost, I. Fleming), Pergamon, Oxford, **1991**.
- [2] For the reactions of fluoro- or polyhalo-substituted olefins and acetylenes, see: a) R. D. Chambers, *Fluorine in Organic Chemistry*, Wiley, New York, **1973**; b) S. Tanimoto, R. Taniyasu, T. Takahashi, T. Miyake, M. Okano, *Bull. Chem. Soc. Jpn.* **1976**, *49*, 1931–1936; c) A. S. Kende, P. Fludzinski, J. H. Hill, W. Swenson, J. Clardy, *J. Am. Chem. Soc.* **1984**, *106*, 3551–3562; d) A. Moyano, F. Charbonnier, A. E. Greene, *J. Org. Chem.* **1987**, *52*, 2919–2922; e) L. M. Geary, P. G. Hultin, *J. Org. Chem.* **2010**, *75*, 6354–6371.
- [3] For brief survey of synthetic utility of haloacetylenes, see: a) S. I. Miller, J. I. Dickstein, *Acc. Chem. Res.* **1976**, *9*, 358–363; b) A. Trofimov, N. Chernyak, V. Gevorgyan, *J. Am. Chem. Soc.* **2008**, *130*, 13538–13539.
- [4] For reviews on nucleophilic addition to haloacetylenes, see: a) R. D. Chambers, S. R. James, *Comprehensive Organic Chemistry, Vol. 1* (Eds.: D. Barton, W. D. Ollis, J. F. Stoddart), Pergamon, Oxford, **1979**, pp. 557–560; b) G. Himbert in *Methoden der Organischen Chemie (Houben-Weyl)*, Vol. E15, Part 3 (Eds.: H. Kropf, E. Schaumann), Georg Thieme, Stuttgart, **1993**, pp. 3319–3329.
- [5] For nucleophilic addition to haloacetylenes, see the following. Hydride: a) G. Zweifel, W. Lewis, H. P. On, *J. Am. Chem. Soc.* **1979**, *101*, 5101–5102; Halides: b) R. Tanaka, S.-Q. Zhèng, K. Kawaguchi, T. Tanaka, *J. Chem. Soc. Perkin Trans. 2* **1980**, 1714–1720; c) Z. Chen, H. Jiang, Y. Li, C. Qi, *Chem. Commun.* **2010**, 46, 8049–8051; Sulfonamide: d) M. Yamagishi, K. Nishigai, T. Hata, H. Urabe, *Org. Lett.* **2011**, *13*, 4873–4875; Imidazoline and Imidazole: e) M. Yamagishi, J. Okazaki, K. Nishigai, T. Hata, H. Urabe, *Org. Lett.* **2012**, *14*, 34–37; Addition of thiols was mentioned as an intermediate in the reaction of 1,1-dibromoolefins and a few nucleophiles: f) H. Xu, S. Gu, W. Chen, D. Li, J. Dou, *J. Org. Chem.* **2011**, *76*, 2448–2458.
- [6] For intramolecular nucleophilic addition to haloacetylene, see the following. With amino group: a) V. N. Elokhina, T. I. Yaroshenko, A. S. Nakhmanovich, A. I. Albanov, *Russ. J. Org. Chem.* **2006**, *42*, 1866–1867; b) V. N. Elokhina, A. S. Nakhmanovich, L. I. Larina, T. I. Yaroshenko, S. V. Amosova, *Russ. J. Org. Chem.* **2009**, *45*, 226–228; With hydroxy group: c) D. Grandjean, P. Pale, J. Chucho, *Tetrahedron Lett.* **1992**, *33*, 4905–4908; d) A. S. Nakhmanovich, V. N. Elokhina, L. I. Larina, E. V. Abramova, V. A. Lopyrev, *Russ. J. Gen. Chem.* **2005**, *75*, 437–439; e) Z. Miao, M. Xu, B. Hoffmann, B. Bernert, A. Vasella, *Helv. Chim. Acta* **2005**, *88*, 1885–1912.
- [7] Transition-metal-catalyzed addition of nucleophiles to haloacetylenes has been reported. Under copper catalysis, see: a) B. Das, G. C. Reddy, P. Balasubramanyam, N. Salvanna, *Synthesis* **2011**, 816–820; b) G. A. Burley, D. L. Davies, G. A. Griffith, M. Lee, K. Singh, *J. Org. Chem.* **2010**, *75*, 980–983; For a reaction that is proposed to involve silver-catalyzed addition, see: c) Z. Chen, J. Li, H. Jiang, S. Zhu, Y. Li, C. Qi, *Org. Lett.* **2010**, *12*, 3262–3265; For intramolecular additions under gold catalysis, see: d) A. K. Buzas, F. M. Istrate, F. Gagosz, *Tetrahedron* **2009**, *65*, 1889–1901; e) H. Harkat, A. Y. Dembelé, J.-M. Weibel, A. Blanc, P. Pale, *Tetrahedron* **2009**, *65*, 1871–1879; f) A. Buzas, F. Gagosz, *Org. Lett.* **2006**, *8*, 515–518; For reviews on relevant metal-mediated hydroamination of acetylenes, see: g) F. Alonso, I. P. Beletskaya, M. Yus, *Chem. Rev.* **2004**, *104*, 3079–3159; h) F. Pohlki, S. Doye, *Chem. Soc. Rev.* **2003**, *32*, 104–114.
- [8] For our reports on the synthetic application of sulfonamides, see: a) Y. Fukudome, H. Naito, T. Hata, H. Urabe, *J. Am. Chem. Soc.* **2008**, *130*, 1820–1821; b) H. Naito, T. Hata, H. Urabe, *Org. Lett.* **2010**, *12*, 1228–1230; c) Y. Kato, D. H. Yen, Y. Fukudome, T. Hata, H. Urabe, *Org. Lett.* **2010**, *12*, 4137–4139.
- [9] For recent reviews on the indole synthesis, see: a) J. Barluenga, C. Valdés, *Modern Heterocyclic Chemistry, Vol. 1* (Eds.: J. Alvarez-Builla, J. J. Vaquero, J. Barluenga), Wiley-VCH, Weinheim, **2011**, pp. 377–531; b) K. Krüger (née Alex), A. Tillack, M. Beller, *Adv. Synth. Catal.* **2008**, *350*, 2153–2167; c) G. R. Humphrey, J. T. Kuethe, *Chem. Rev.* **2006**, *106*, 2875–2911; d) L. Ackermann, *Synlett* **2007**, 507–526; For recent reports on indole synthesis, see: e) L. Ackermann, A. Althammer, P. Mayer, *Synthesis* **2009**, 3493–3503; f) M. P. Huestis, L. Chan, D. R. Stuart, K. Fagnou, *Angew. Chem.* **2011**, *123*, 1374–1377; *Angew. Chem. Int. Ed.* **2011**, *50*, 1338–1341; g) D. R. Stuart, P. Alsabeh, M. Kuhn, K. Fagnou, *J. Am. Chem. Soc.* **2010**, *132*, 18326–18339; h) Z. Shi, C. Zhang, S. Li, D. Pan, S. Ding, Y. Cui, N. Jiao, *Angew. Chem.* **2009**, *121*, 4642–4646; *Angew. Chem. Int. Ed.* **2009**, *48*, 4572–4576; i) A. Wetzel, F. Gagosz, *Angew. Chem.* **2011**, *123*, 7492–7496; *Angew. Chem. Int. Ed.* **2011**, *50*, 7354–7358; j) T. Gehrman, J. L. Fillol, S. A. Scholl, H. Wadepohl, L. H. Gade, *Angew. Chem.* **2011**, *123*, 5876–5881; *Angew. Chem. Int. Ed.* **2011**, *50*, 5757–5761; k) S. G. Newman, M. Lautens, *J. Am. Chem. Soc.* **2010**, *132*, 11416–11417; l) S. Würtz, S. Rakshit, J. J. Neumann, T. Dröge, F. Glorius, *Angew. Chem.* **2008**, *120*, 7340–7343; *Angew. Chem. Int. Ed.* **2008**, *47*, 7230–7233.
- [10] For recent reviews on cyclization through C–H activation, see: a) T. W. Lyons, M. S. Sanford, *Chem. Rev.* **2010**, *110*, 1147–1169; b) L. Ackermann, R. Vicente, A. R. Kapdi, *Angew. Chem.* **2009**, *121*, 9976–10011; *Angew. Chem. Int. Ed.* **2009**, *48*, 9792–9826; c) X. Chen, K. M. Engle, D.-H. Wang, J.-Q. Yu, *Angew. Chem.* **2009**, *121*, 5196–5217; *Angew. Chem. Int. Ed.* **2009**, *48*, 5094–5115; d) D. Alberico, M. E. Scott, M. Lautens, *Chem. Rev.* **2007**, *107*, 174–238; For transition-metal-catalyzed coupling of arene and haloalkenes; see: e) M. Yagoubi, A. C. F. Cruz, P. L. Nichols, R. L. Elliott, M. C. Willis, *Angew. Chem.* **2010**, *122*, 8130–8134; *Angew. Chem. Int. Ed.* **2010**, *49*, 7958–7962; f) M. R. Albicker, N. Cramer, *Angew. Chem.* **2009**, *121*, 9303–9306; *Angew. Chem. Int. Ed.* **2009**, *48*, 9139–9142; g) A. C. F. Cruz, N. D. Miller, M. C. Willis, *Org. Lett.* **2007**, *9*, 4391–4393; h) C. C. Hughes, D. Trauner, *Angew. Chem.* **2002**, *114*, 1639–1642; *Angew. Chem. Int. Ed.* **2002**, *41*, 1569–1572; i) L.-C. Campeau, M. Parisien, A. Jean, K. Fagnou, *J. Am. Chem. Soc.* **2006**, *128*, 581–590; For a recent review on mechanistic aspects, see: j) L. Ackermann, *Chem. Rev.* **2011**, *111*, 1315–1345.
- [11] a) D. K. Barange, V. R. Batchu, D. Gorja, V. R. Pattabiraman, L. K. Tatini, J. M. Babu, M. Pal, *Tetrahedron* **2007**, *63*, 1775–1789; b) A. Vidal, J.-C. Madelmont, E. Mounetou, *Synthesis* **2006**, 591–593; c) W. J. Layman, Jr., T. D. Greenwood, A. L. Downey, J. F. Wolfe, *J. Org. Chem.* **2005**, *70*, 9147–9155; d) H. Zinnes, R. A. Comes, F. R. Zuleski, A. N. Caro, J. Shavel, Jr., *J. Org. Chem.* **1965**, *30*, 2241–2246.
- [12] This synthetic advantage of the products for cyclization was firstly addressed in Ref. [5d]. During the course of our study, a similar method for the preparation of aryl-substituted benzofurans from aromatic haloacetylenes was reported: S. Wang, P. Li, L. Yu, L. Wang, *Org. Lett.* **2011**, *13*, 5968–5971, where *aliphatic* haloacetylenes appear sluggish substrates.
- [13] See the Supporting Information.
- [14] A. Yasuhara, T. Sakamoto, *Tetrahedron Lett.* **1998**, *39*, 595–596.
- [15] a) X. Zhang, Z. Sui, *Tetrahedron Lett.* **2003**, *44*, 3071–3073; b) S. Tumkevicius, V. Masevicius, *Synthesis* **2007**, 3815–3820.
- [16] To the contrary, mesylation of a mixture of **4** and **5** could provide the *N*-mesylindoles **4**. P. G. M. Wuts, T. W. Greene, *Greene's Protective Groups in Organic Synthesis, 4th ed.*, Wiley, Hoboken, **2007**, pp. 873–874.
- [17] Amines such as butyl- and diethylamines in place of sulfonamides afforded an intractable mixture of products in the addition to 1-bromo-1-octyne. On the other hand, an aromatic haloacetylene such as bromo(phenyl)acetylene and sulfonamide **2b** afforded the desired addition product **3** in a low yield (ca. 30 %).